

# Combining two-directional synthesis and tandem reactions:† an efficient strategy for the total syntheses of (±)-hippodamine and (±)-*epi*-hippodamine‡

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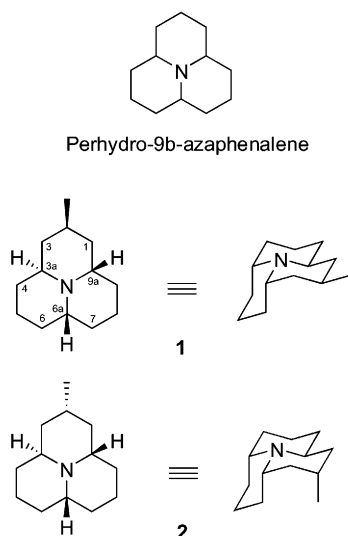
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Two-directional total stereoselective syntheses of (±)-hippodamine and (±)-*epi*-hippodamine, utilising a tandem deprotection/intramolecular double Michael addition sequence as the key step, are presented.

## Introduction

Ladybird beetles (Coleoptera: Coccinellidae) are important predators contributing to the natural control of pest aphid populations and are therefore of considerable interest for insect pest management. However, ladybirds themselves are attacked by a range of natural enemies. General predation on ladybirds by vertebrates such as birds is largely prevented by highly toxic, bitter tasting, defence alkaloids contained in a reflex bleed released when the ladybird is attacked. To date, eight alkaloids of this type have been isolated from coccinellid beetles,<sup>1</sup> all of them being formally derivatives of perhydro-9b-azaphenalene (or dodecahydro-pyrido[2,1,6-*de*]quinolizine) (Fig. 1).



**Fig. 1** Structures of hippodamine (**1**) and *epi*-hippodamine (**2**), derivatives of perhydro-9b-azaphenalene.

Another group of natural enemies, parasitic insects, can cause substantial reductions in populations of ladybird species. Recent research<sup>2</sup> has shown that the parasites locate the ladybirds through perception of certain defence alkaloids that they emit. If ladybirds are to be used effectively in insect pest control then their parasites must be controlled as well. The significant

attraction of parasitic insects to the ladybird alkaloids suggests that there is potential for development of control strategies for this particular natural enemy.

In addition to their bitter taste, the coccinellid alkaloids also seem to interfere with the nervous systems of higher animals. Certain derivatives of coccinellid alkaloids have recently been shown<sup>3</sup> to be potent receptor antagonists of the neurotransmitter serotonin. This pharmacological property makes these compounds attractive for the treatment of a number of gastrointestinal and brain disorders. One of the most intriguing properties is the ability of these compounds to prevent behavioural consequences of the withdrawal of drugs such as nicotine. For these reasons much attention has been paid to developing syntheses of the coccinellid defence alkaloids.<sup>1</sup>

Hippodamine (**1**) is a naturally occurring alkaloid isolated from a ladybird beetle *Hippodamia convergens* by Tursch and co-workers in 1972.<sup>4</sup> The structure of hippodamine (**1**) was established two years later by the same group<sup>5</sup> on the basis of a single-crystal X-ray diffraction experiment (Fig. 1). *epi*-Hippodamine (**2**) is a non-natural isomer with an axial C(2) methyl group. Both hippodamine (**1**)<sup>6,7,8</sup> and *epi*-hippodamine (**2**)<sup>9</sup> have been the subjects of total synthesis.

In our previous short communication<sup>10</sup> we reported on the formation of quinolizidine **8**. Herein, we give a full account of a concise synthetic strategy to the title azaphenalene alkaloids **1** and **2** using a two-directional approach to a common key intermediate, with a tandem deprotection/intramolecular double Michael addition (IDM) sequence as the key step.

## Results and discussion

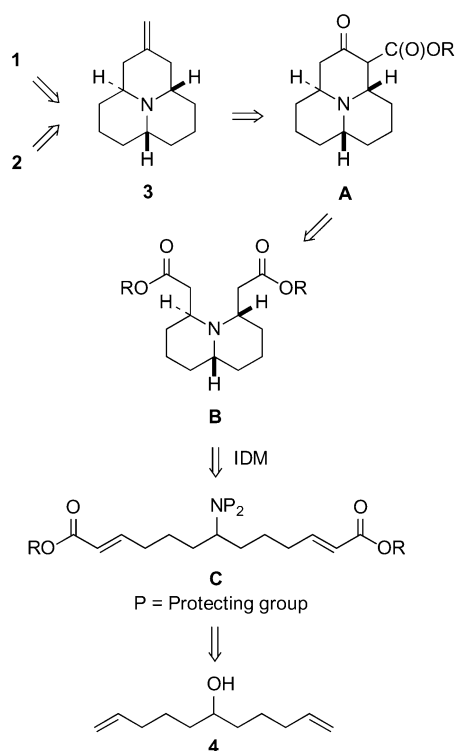
### Retrosynthesis

Assembling the tricyclic skeleton and controlling the stereochemistry of the title azaphenalene alkaloids in a minimum of synthetic steps represents a formidable synthetic challenge. Our plan was to establish the axial or equatorial C(2) methyl substituents of the title compounds by manipulation of the exocyclic double bond in **3** (Fig. 2). The olefin **3** could be formed by de-alkoxycarbonylation of **A** and a subsequent Wittig olefination.

Dieckmann condensation of diester **B** would in turn furnish the third ring of the tricyclic structure. The formation of *trans*-4,6-disubstituted quinolizidine derivative **B** represents the key step in the synthetic plan. It was assumed that the quinolizidine derivative **B** could be accessed *via* an intramolecular double Michael addition of the acyclic intermediate **C** and the desired *trans*-relative configuration of the substituents in **B** could be installed at the same time. The acyclic intermediate **C** is a symmetric molecule opening up the possibility of two-directional

† Part 5. For Part 4 in the series see: L. G. Arini, P. Szeto, D. L. Hughes and R. A. Stockman, *Tetrahedron Lett.*, 2004, **45**, 8371.

‡ Electronic supplementary information (ESI) available: <sup>13</sup>C NMR spectra for compounds **1**, **2**, **3**, **7**, **10**, **13**, **14**, **18**, **22**, **23**, and **25**. X-Ray data for compounds **22** and **25**. See <http://www.rsc.org/suppdata/ob/b4/b413052a/>

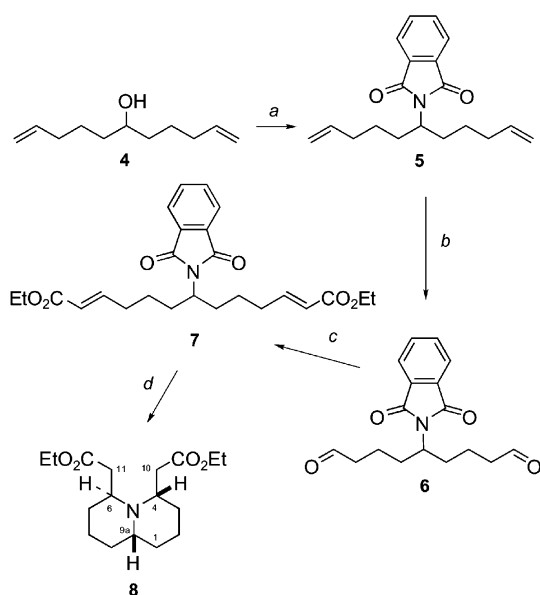


**Fig. 2** Retrosynthesis for (±)-hippodamine (**1**) and (±)-*epi*-hippodamine (**2**).

synthesis. It was thought that **C** could be obtained from undeca-1,10-dien-6-ol (**4**) by firstly installing and protecting the required amino group, and secondly by oxidative cleavage of both double bonds and by a subsequent olefination. Finally, undeca-1,10-dien-6-ol (**4**) can be obtained from 5-bromo-1-pentene by a known procedure.

### Synthesis of common intermediate **3**

Diester **7** (Scheme 1) was prepared using the principles of two-directional synthesis. The symmetrical alcohol (**4**) was prepared in 89% yield following the method of Kitching *et al.*<sup>11</sup> Mitsunobu substitution<sup>12</sup> of alcohol **4** with phthalimide was then carried



**Scheme 1** (a) Phthalimide,  $\text{PPh}_3$ , DIAD, THF (97%); (b) Method A:  $\text{OsO}_4$ ,  $\text{NaIO}_4$ , Method B:  $\text{RuCl}_3$ ,  $\text{NaIO}_4$ , Method C: (1)  $\text{O}_3$ ,  $\text{CH}_2\text{Cl}_2$ ,  $-78^\circ\text{C}$ , (2)  $\text{Me}_2\text{S}$ ,  $\text{PPh}_3$ , rt; (c)  $\text{Et}_3\text{O}(\text{O})\text{PCH}_2\text{CO}_2\text{Et}$ ,  $\text{NaH}$ , THF (over 2 steps Method A: 75%, Method B: 71%, Method C: 45%); (d) (1)  $\text{NaBH}_4$  (1.4 eq.),  $i\text{-PrOH}/\text{H}_2\text{O}$  (6.2:1), (2)  $\text{AcOH}$  *glac.* (71%).

out to give diene **5** in 97% yield after column chromatography.

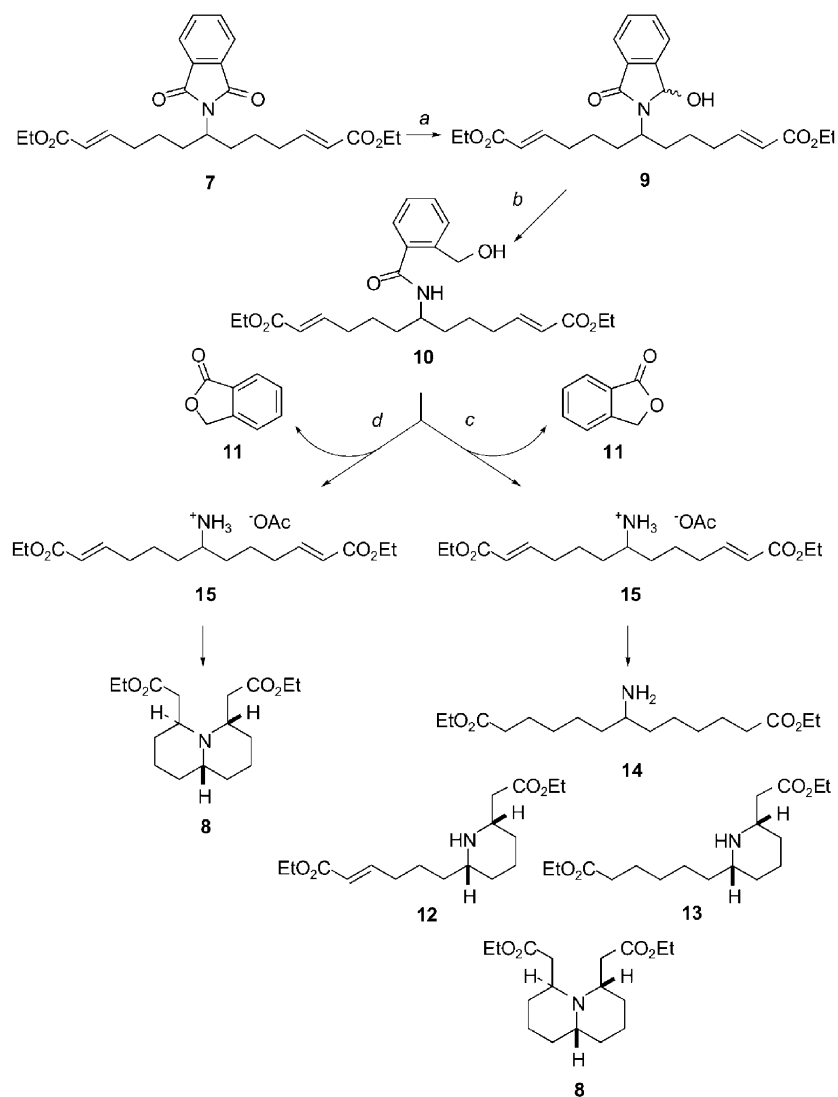
Oxidative cleavage of the terminal alkenes in **5** was evaluated using three different methods: treatment of **5** with sodium periodate (6 eq.) and a catalytic amount of osmium tetroxide (2 mol%) in 3 : 1 THF–water<sup>13</sup> (Method A); ruthenium(III) chloride (4.1 mol%) and sodium periodate (4 eq.) in 6 : 1  $\text{CH}_3\text{CN}$ –water<sup>14</sup> (Method B); and ozone in dichloromethane and subsequent treatment with  $\text{Me}_2\text{S}$  and  $\text{PPh}_3$ <sup>15</sup> (Method C). The resulting crude dialdehyde **6** was doubly homologated under standard Wadsworth–Emmons conditions<sup>16</sup> to yield acyclic diester **7** in 75% (Method A), 71% (Method B) or 45% (Method C) yield respectively over two steps from diene **5** as a single *E,E*-stereoisomer. The ruthenium(III) chloride catalysed oxidation (Method B) gave varying results which was later put down to the varying quality of the catalyst batches. Ozonolysis (Method C) proved to be difficult to perform on a larger scale. The osmium tetroxide catalysed oxidative cleavage (Method A) gave excellent yields on a small scale (up to 92%).<sup>10</sup> Although the yield dropped to some extent (75%) when performed on a large scale (>10 g), Method A was the method of choice for preparation of **7** as the results were easily reproducible.

With the successful formation of protected amine **7**, the stage was now set for the key tandem deprotection/double intramolecular Michael addition, which we surmised should give our desired *trans*-4,6-disubstituted quinolizidine **8**. Following Ganem's protocol,<sup>17</sup> the phthalimide group was removed by selective hydride reduction (Scheme 2) of the alkyl succinimide **7** to furnish an *ortho*-hydroxymethyl benzamide **10** (probably *via* hemiaminal **9**). Addition of acetic acid to the reaction solution (*c* in Scheme 2) promotes lactonisation of the *ortho*-hydroxymethyl benzamide **10** to form phthalide **11** with concomitant release of the primary amine, or its hydroacetate **15** respectively. The hydroacetate must however, at least to some extent, equilibrate with the free amine representing the reactive species in the desired Michael addition. Under the acidic conditions the amine undergoes an intramolecular Michael addition to give the product of the first cyclisation **12**, which could be isolated from the reaction mixture with **13**. The second cyclisation can follow to give the desired quinolizidine derivative **8**. However, under the initial unoptimised conditions (*c* in Scheme 2) other side-reactions were noted: the side chain double bond of **12** was partially reduced (to give **13**) and so were both double bonds in **15** (to give **14**). As a consequence the Michael addition could not occur and the yield of **8** was low. We suspected that sodium triacetoxyborohydride<sup>18</sup> formed *in situ* from sodium borohydride and acetic acid might be responsible for this undesired over-reduction.

To overcome this problem any surplus of the hydride was quenched by addition of an aliquot of acetone prior to addition of the acetic acid (*d* in Scheme 2). When all remaining hydride was removed acetic acid was added promoting lactonisation as in the previous case and liberating the amine hydroacetate **15**. Under the acidic conditions the acyclic unsaturated amine undergoes the intramolecular double Michael addition to give the desired quinolizidine derivative **8** in high yield.

The 2,6-disubstituted piperidine **13** showed a *cis*-relative configuration of the ring substituents. The *cis*-configuration was assigned on the basis of NMR spectra and from analogy with spectra of a *cis/trans* pair of another 2,6-dialkyl piperidine compound.<sup>19</sup> The structure of **12** was assigned from analogy with **13**. From this observation it is possible to conclude that the first Michael addition of **15** leads to a thermodynamically more stable *cis*-2,6-dialkyl piperidine **12**. The stereochemistry of the subsequent Michael addition of **12** is then controlled by the stereochemistry of the first ring. As a result the formation of *trans*-4,6-disubstituted quinolizidine **8** as a single isomer is observed.

Optimal conditions for the tandem deprotection/double cyclisation were as follows: *N*-alkyl phthalimide **7** (Scheme 1) was



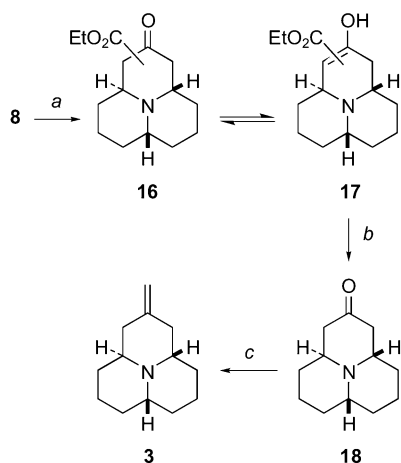
**Scheme 2** Proposed mechanism of the tandem deprotection/intramolecular double Michael addition sequence: (a) NaBH<sub>4</sub> (0.25 eq.), *i*-PrOH–H<sub>2</sub>O (6.2 : 1) (product not isolated); (b) NaBH<sub>4</sub> (0.25 eq.), *i*-PrOH–H<sub>2</sub>O (6.2 : 1) (67%); (c) (1) AcOH glac., (2) K<sub>2</sub>CO<sub>3</sub>, benzene (mixture of products **8**, **11**, **12**, **13** and **14**); (d) (1) acetone, 12 h, (2) AcOH glac., (3) K<sub>2</sub>CO<sub>3</sub>, benzene (71%).

subjected to sodium borohydride (1.4 eq.) in 6.2 : 1 isopropanol–water mixture at 25 °C for 24 hours.<sup>17</sup> After this time acetone (3.6 eq.) was added and stirring was continued overnight to quench any remaining hydride. Then glacial acetic acid (30 eq.) was added dropwise at 0 °C, and the reaction mixture was heated to 80 °C (bath temperature) for a further 24 hours. An aqueous work-up gave a crude hydroacetate of quinolizidine **8**. The free base **8** was liberated by treating the hydroacetate with K<sub>2</sub>CO<sub>3</sub> in benzene. Final purification on activity IV neutral alumina produced pure quinolizidine **8** as a single ( $\pm$ )-*trans*-isomer (according to NMR) in 71% yield.

The *trans* relative configuration of substituents in positions C(4) and C(6) of the quinolizidine **8** was determined by NMR. Protons C(4)-H and C(6)-H (Scheme 1) situated on the  $\alpha$ -carbon atoms to the nitrogen atom show different behaviour in the <sup>1</sup>H NMR spectrum. Proton C(6)-H (2.78–2.72, m) is more shielded than C(4)-H (3.83–3.80, m), the deshielding being about 1 ppm. This can only be explained by an antiperiplanar spatial arrangement assumed by C(6)-H and the nitrogen lone pair. Consequently C(6)-H proton must be in an axial and C(4)-H in an equatorial position with respect to the *trans*-quinolizidine moiety. Moreover, in the <sup>13</sup>C NMR spectrum, the axial methylene group C(4)-CH<sub>2</sub> (26.3, t) of the side chain, is more shielded than the respective equatorial C(6)-CH<sub>2</sub> (38.5, t), the deshielding being 12.2 ppm. This can be explained by

a double  $\gamma$ -*gauche* effect<sup>20</sup> experienced by the axial C(4)-CH<sub>2</sub> methylene group from C(2) and C(9a). No such effect is suffered by the respective equatorial C(6)-CH<sub>2</sub> group. In addition, in the <sup>13</sup>C NMR, 16 signals were detected rather than the 9 that would be expected for the more symmetric *cis*-isomer of the quinolizidine **8**.

We had planned that the formation of the third ring of the perhydro-9b-azaphenalene skeleton would be achieved using a Dieckmann condensation.<sup>9,21</sup> To this effect, the 4,6-disubstituted quinolizidine derivative **8** was subjected to a variety of bases and reaction conditions. Two methods were found to give the desired cyclisation product. When **8** was treated with LDA in THF at –78 °C (Scheme 3) a mixture of  $\beta$ -ketoester **16** and its corresponding enol form **17** was obtained. Initially when the reaction mixture was directly subjected to an aqueous work-up the resulting yields were very low. This was later put down to the low stability of the cyclic products **16** and **17** towards aqueous basic solutions. This problem was circumvented by removal of THF and by taking the residue into benzene prior to quenching of the reaction by an aqueous solution. This method eventually gave a reasonable yield (48%) of the desired products **16** and **17**. To further improve the yield we explored the alternative of performing the Dieckmann condensation in a lipophilic solvent. When **8** was treated with *t*BuOK in refluxing benzene (Scheme 3) under continuous azeotropic removal of EtOH using a



**Scheme 3** (a) Method A: *t*BuOK, Li, benzene, azeotropic removal of EtOH (84%), Method B: LDA, THF,  $-78^{\circ}\text{C}$  (48%); (b) (1) HCl, Et<sub>2</sub>O, (2) LiCl, wet DMF, reflux (66%); (c) CH<sub>3</sub>PPh<sub>3</sub>Br, *n*BuLi, THF (87%).

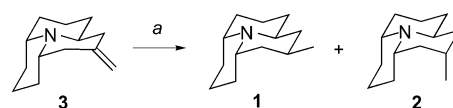
Dean–Stark trap with lithium in the trap to remove EtOH, a mixture of  $\beta$ -ketoester **16** and its corresponding enol form **17** was obtained in a very good yield (84%). The Dieckmann condensation gave a mixture of regioisomeric  $\beta$ -ketoesters **16** differing in the position of the ester moiety on either C(1) or C(3). Moreover, both regioisomers of **16** are in equilibrium with their enol forms **17**. However, the following step, a de-ethoxycarbonylation of all the possible isomers, would lead to the ketone **18** only and consequently we did not purify and identify the components of the isomeric mixture any further.

The de-ethoxycarbonylation of the isomeric mixture of **16** and **17** gave initially only limited yields of **18** when the mixture was directly subjected to LiCl in wet refluxing DMF. It appeared that only the  $\beta$ -ketoester **16** was able to undergo the desired decarboxylation. The enol form **17** was either recovered or it decomposed after a prolonged reaction time. This problem was eventually solved by exposing the isomeric mixture **16** and **17** to HCl in Et<sub>2</sub>O prior to the treatment with LiCl in wet refluxing DMF. This method gave the desired ketone **18** in 66% yield after purification.

Wittig olefination<sup>7</sup> of the ketone **18** using methyltriphenylphosphonium bromide in THF and *n*BuLi gave low yields of the desired product **3** when the olefination was performed between  $-78^{\circ}\text{C}$  and room temperature. However, a short reflux of the reaction mixture gave the desired product **3** in a good yield (87%) after purification.

### ( $\pm$ )-Hippodamine (**1**)

In order to achieve our objective the exocyclic double bond of the olefin **3** had to be hydrogenated in a stereoselective way. In fact the olefin **3** would have to approach the catalyst surface *via* its more hindered face in order to obtain the required equatorial methyl on C(2) (Scheme 4) which of course is less likely than the opposite case. Indeed, when a catalytic hydrogenation using 10% palladium on carbon in absolute methanol was performed (Table 1, Entry 1) a mixture of ( $\pm$ )-hippodamine (**1**) and ( $\pm$ )-*epi*-hippodamine (**2**) was obtained in a high combined yield (99%), but in a ratio **1** : **2** of 1 : 1.4, thus supporting the theory that the olefin **3** is much more likely to approach the catalyst surface *via* its less hindered face, with the hydrogenation resulting in an axial methyl substituent.



**Scheme 4** (a) Pd–C 10%, H<sub>2</sub> (normal pressure), solvent, additive (for details see Table 1).

In order to prepare the target compound ( $\pm$ )-hippodamine (**1**) the selectivity of the hydrogenation had to be reversed. In order to achieve this we decided to make use of the nitrogen lone pair in **3**. Reacting the nitrogen lone pair with a convenient reagent would hopefully block one face of the rigid olefin **3** thus allowing the resulting product (a salt or a complex) to approach the catalyst *via* the desired opposite face only.

When the free base of olefin **3** was reacted at room temperature with 0.5 equivalent of oxalic acid in MeOH a dimeric salt was formed immediately (Table 1, Entry 2). This salt was subjected to catalytic hydrogenation at room temperature and atmospheric pressure of hydrogen. The hydrogenation was monitored using TLC. When the conversion was complete, the mixture was filtered, evaporated and the residue treated with K<sub>2</sub>CO<sub>3</sub> in benzene to liberate the free base. After purification on activity IV neutral alumina a mixture of ( $\pm$ )-hippodamine (**1**) and ( $\pm$ )-*epi*-hippodamine (**2**) was obtained in a high combined yield (98%). This time the ratio **1** : **2** amounted to 1.2 : 1 in favour of the desired ( $\pm$ )-hippodamine (**1**). Although not a significant advance, this experiment did show that it was possible to reverse the stereoselectivity of the hydrogenation.

To further increase the selectivity of the hydrogenation for ( $\pm$ )-hippodamine (**1**) a range of salts and boron complexes were screened for their behaviour in hydrogenation (Table 1). The best results were obtained when a salt of **3** with mesitylenesulfonic acid (Table 1, entries 7–9) was hydrogenated. A variety of solvents and salt formation conditions were tested to show that

**Table 1** Stereoselectivity of hydrogenation of olefin **3** and its salts or complexes

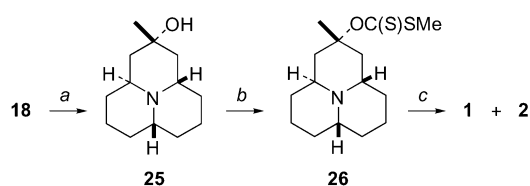
Entry	Additive	Solvent	Time/days	<i>T</i> / $^{\circ}\text{C}$ ; solvent <sup>b</sup>	Combined yield (%)	Ratio <b>1</b> : <b>2</b> <sup>a</sup>	ds <sup>c</sup> (%)
1	—	MeOH	1	rt	99	1 : 1.4	17.4 ( <b>2</b> )
2	Oxalic acid (0.5 eq.)	MeOH	1	rt	98	1.2 : 1	10.0 ( <b>1</b> )
3	Diphenylacetic acid	Et <sub>2</sub> O	1	$-20$	92	2.0 : 1	34.0 ( <b>1</b> )
4	(1 <i>S</i> )-(+)-10-Camphorsulfonic acid	MeOH	1	rt; CH <sub>2</sub> Cl <sub>2</sub>	99	1.8 : 1	28.0 ( <b>1</b> )
5	(1 <i>S</i> )-(+)-10-Camphorsulfonic acid	AcOEt	2	rt; CH <sub>2</sub> Cl <sub>2</sub>	97	2.7 : 1	46.0 ( <b>1</b> )
6	(1 <i>S</i> )-(+)-10-Camphorsulfonic acid	THF	3	rt; CH <sub>2</sub> Cl <sub>2</sub>	97	3.3 : 1	54.0 ( <b>1</b> )
7	Mesitylenesulfonic acid	THF	2	rt	93	3.5 : 1	56.0 ( <b>1</b> )
8	Mesitylenesulfonic acid	THF	2 at $-20^{\circ}\text{C}$	$-20$	95	2.8 : 1	48.0 ( <b>1</b> )
9	Mesitylenesulfonic acid	Et <sub>2</sub> O–THF 4 : 1	2	rt; Et <sub>2</sub> O	99	2.0 : 1	34.0 ( <b>1</b> )
10	2,4,6-Triisopropylbenzenesulfonic acid	Et <sub>2</sub> O	1	rt	98	1.8 : 1	28.0 ( <b>1</b> )
11	9-Cyclohexyl 9-borabicyclo[3.3.1]nonane	MeOH	1	rt	79	1 : 1.6	24.0 ( <b>2</b> )

<sup>a</sup> The ratio was determined from relative intensities of the C(2) methyl substituent signals in 400 MHz <sup>1</sup>H NMR spectra. <sup>b</sup> Temperature and solvent (if other than the hydrogenation solvent) of the salt formation. <sup>c</sup> Diastereoselectivity in favour of either (**1**) or (**2**).

THF at room temperature gave a **1** : **2** ratio of 3.5 : 1 (56% ds) in a very high combined yield (93%). This mixture was then subjected to multiple column chromatography on neutral aluminium oxide activity IV affording isomerically pure hippodamine in 37% yield.

In summary, we have achieved the synthesis of ( $\pm$ )-hippodamine (**1**) in 8 synthetic steps in an overall yield of 7.2%, using the concept of combining two-directional synthesis and tandem reactions.

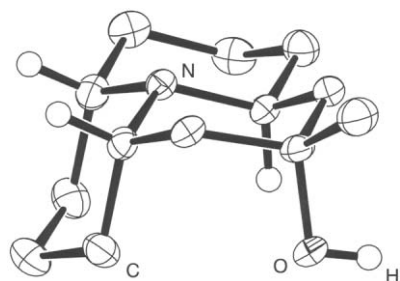
In an attempt to further improve the control over the formation of the equatorial methyl substituent in **1** we planned to perform a deoxygenation of tertiary alcohol **25** (Scheme 5) via its xanthate **26**. The Barton–McCombie deoxygenation of xanthate ester **26** should in theory proceed via a planar radical species. We assumed that the nitrogen lone pair might donate into the half-filled p-orbital and force the unpaired electron into the opposite face. As a consequence, the hydrogen radical should only react from the required face which would then lead to the desired equatorial methyl substituent of **1**.



**Scheme 5** (a) (1) MeLi, toluene, 0 °C, (2) crystallization from Et<sub>2</sub>O (61%); (b) (1) *n*BuLi, THF, –78 °C, (2) CS<sub>2</sub>, rt, (3) MeI (42%); (c) Bu<sub>3</sub>SnH, AIBN, toluene, reflux (46%).

The tricyclic ketone **18** was transformed into tertiary alcohol **25** (Scheme 5). Interestingly, when the ketone was treated with MeMgBr in either polar or lipophilic aprotic solvents, no or only very limited formation of the desired alcohol was observed. This was most likely due to competing enol formation. When the ketone **18** was treated with MeLi in THF the outcome was very similar. When however toluene was used as a solvent and the reaction was performed at 0 °C the desired tertiary alcohol was formed and isolated in 61% yield after crystallization from diethyl ether.

A single crystal X-ray diffraction experiment of **25**<sup>22</sup> (Fig. 3) confirmed isolation of a single isomer only with axial hydroxyl and equatorial methyl substituents. It also confirmed the expected *trans,cis,trans*-stereochemistry of the tricyclic system. The ability of this alcohol to crystallize was greatly enhanced by formation of a hydrogen bond between the hydroxy group and the nitrogen lone pair of a neighbouring molecule.



**Fig. 3** Crystal structure of alcohol **25**.

The alcohol **25** was then reacted with 1,1'-thiocarbonyl-diimidazole in THF in an attempt to prepare the corresponding xanthate. This method failed, probably due to steric hindrance. For this reason the alcohol **25** was first deprotonated using *n*BuLi in THF at –78 °C. The corresponding alkoxide was then treated with CS<sub>2</sub> and the reaction mixture was allowed to reach room temperature. The reaction was then quenched using MeI to give the desired xanthate **26**. This compound

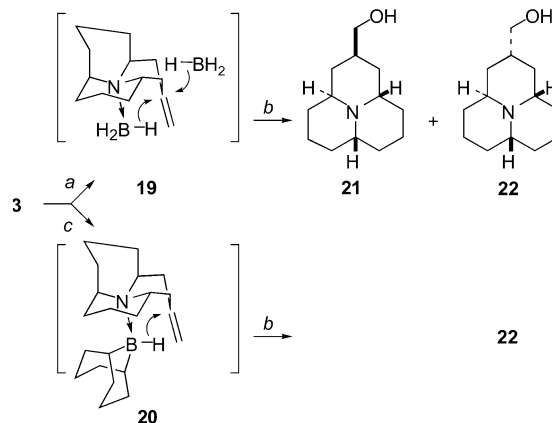
was not very stable and consequently it was used after a short filtration through a column of neutral alumina without further purification. Finally, xanthate **26** was subjected to the Barton–McCombie deoxygenation conditions giving 46% combined yield of a mixture of **1** and **2** in a ratio of 2.3 : 1.

The predicted donation of the nitrogen lone pair in the planar radical intermediate of the Barton–McCombie deoxygenation is very likely operative but the control over the formation of the desired equatorial methyl substituent was only limited, most likely due to the high reaction temperature employed. The desired ( $\pm$ )-hippodamine (**1**) was formed in 39.4% ds.

### ( $\pm$ )-*epi*-Hippodamine (**2**)

With the successful completion of our synthesis of ( $\pm$ )-hippodamine (**1**) we also wished to prepare the non-natural epimer ( $\pm$ )-*epi*-hippodamine (**2**) as it was thought that it might provide insight into the influence of stereochemistry on the biological activity of these compounds.

We surmised that we could use the nitrogen lone pair to deliver a borohydride reagent selectively to one face of the exocyclic olefin **3**. Hydroboration of the exocyclic double bond in **3** (Scheme 6) should lead to an *anti*-Markovnikov primary alcohol. The hydroboration however can proceed from two possible faces leading to either the equatorial **21** or the axial alcohol **22**. The tertiary amine moiety in **3** makes this molecule a good Lewis base. Such a base should be capable of quickly replacing ligands of ether type (weaker Lewis bases) in borane etherate complexes and form complexes such as **19** and **20** at low temperatures. At elevated temperatures the hydroboration should hopefully then occur in an intramolecular fashion.



**Scheme 6** (a) BH<sub>3</sub>·THF, THF, –78 °C; (b) (1) BH<sub>3</sub>·THF, THF, rt to reflux, (2) H<sub>2</sub>O<sub>2</sub>, NaOH, H<sub>2</sub>O (**21** + **22** 90%); (c) 9-BBN·THF, THF, –78 °C; (d) (1) 9-BBN·THF, THF, rt to reflux, (2) H<sub>2</sub>O<sub>2</sub>, NaOH, H<sub>2</sub>O (**22** only 84%).

Olefin **3** was treated with borane tetrahydrofuran complex in absolute THF at –78 °C for a short time period. The mixture was then allowed to warm up to room temperature and finally the mixture was heated at reflux. Oxidative work-up under basic conditions furnished a mixture of alcohols **21** and **22** in a ratio 1 : 4.3, with the axial alcohol being favoured, to our gratification. In an attempt to further increase the selectivity olefin **3** was treated with 9-borabicyclo[3.3.1]nonane tetrahydrofuran complex. The hydroboration furnished exclusively the axial alcohol **22** in a high yield (84%) after crystallization from Et<sub>2</sub>O.

A single crystal X-ray diffraction experiment of the alcohol **22**<sup>23</sup> (Fig. 4) confirmed the axial position of the hydroxymethylene moiety. Similarly to the alcohol **25** the ability of this compound to crystallize was greatly enhanced by formation of hydrogen bonds between the hydroxy group and the nitrogen lone pair of a neighbouring molecule.

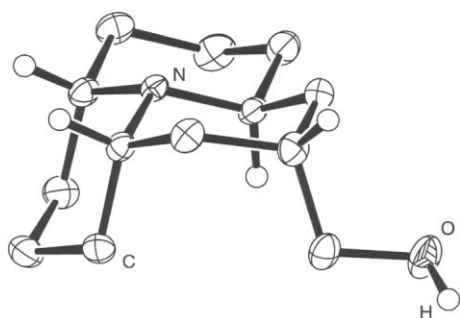
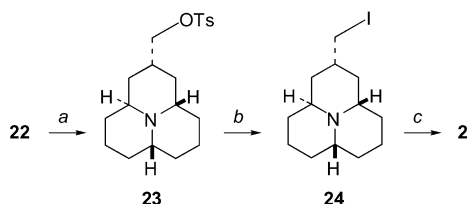


Fig. 4 Crystal structure of alcohol 22.



**Scheme 7** (a) Method A: TsCl, Et<sub>3</sub>N, Me<sub>3</sub>N·HCl, CH<sub>2</sub>Cl<sub>2</sub> (25%), Method B: (1) *n*BuLi, THF, (2) TsCl, THF (95%); (b) NaI, 2-butanone (usually not isolated); (c) (1) Zn activated, AcOH glac., (2) filtration, evaporation, (3) K<sub>2</sub>CO<sub>3</sub>, benzene (83%).

Removal of the oxygen functionality (Scheme 7) by conversion to the iodide **24** via tosylate **23**<sup>24</sup> followed by reduction with zinc<sup>25</sup> in refluxing acetic acid gave the desired ( $\pm$ )-*epi*-hippodamine (**2**) in a good yield (83% from **23**) after purification. Attempts to directly reduce (LiAlH<sub>4</sub>, LiAlH<sub>4</sub>-AlCl<sub>3</sub> to give AlH<sub>3</sub>, superhydride LiB(C<sub>2</sub>H<sub>5</sub>)<sub>3</sub>H) the tosylate **23** to the corresponding alkane **2** failed. Similarly all attempts to proto-deborylate the putative alkyl borane intermediate formed on hydroboration of **3** also failed.

In summary, we have managed to synthesise isomerically pure ( $\pm$ )-*epi*-hippodamine (**2**) in 11 synthetic steps in an overall yield of 17.0%.

Studies on syntheses of the other members of the perhydro-9b-azaphenanthrene class of alkaloids and their application in insect pest management are ongoing in these laboratories, and will be reported in due course.

## Experimental

### General methods

All chemicals were purchased as reagent grade and used without further purification. Tetrahydrofuran was freshly distilled from sodium diphenyl ketyl just prior to use. Dichloromethane was freshly distilled over CaH<sub>2</sub>. Other solvents for experiments were purified by the usual methods or purchased from a commercial source. If not stated otherwise TLC was performed on precoated silica plates containing a fluorescence indicator. Silica gel (63–200  $\mu$ m) was used for analytical and preparative TLC (PLC) and for column chromatography, unless otherwise noted. Compounds were visualised under UV (254 nm), by heating after dipping in a solution of alkaline KMnO<sub>4</sub>, or by exposing to iodine vapours. NMR spectra were recorded on 300 or 400 MHz instruments. Chemical shifts are reported in ppm with respect to TMS. Coupling constants are reported in Hz. Multiplicity of signals in <sup>13</sup>C NMR spectra was determined from DEPT or gHSQC spectra. IR spectra of neat (unless otherwise noted) compounds were expressed as wave numbers (cm<sup>-1</sup>). The Mass Spectrometry Service Centre at the University of Wales, Swansea performed the low- and high-resolution mass spectra. Melting points were taken on a micro melting point apparatus and are uncorrected. Water- and air-sensitive reactions were

routinely carried out under argon atmosphere. Compound **4** was synthesized according to literature procedure.<sup>11</sup>

### 2-(1'-Pent-4"-enyl-hex-5'-enyl)-isoindole-1,3-dione (**5**)

A 2-necked 1000 mL flask equipped with addition funnel and magnetic stirrer was charged with triphenyl phosphine (42.9 g, 163.4 mmol), phthalimide (24.0 g, 163.4 mmol), anhydrous THF (700 mL) and undeca-1,10-dien-6-ol (**4**) (25 g, 148.6 mmol). The mixture was cooled to -20 °C under stirring and argon atmosphere. Diisopropyl azodicarboxylate (33.0 g, 32.4 mL, 163.4 mmol) in anhydrous THF (100 mL) was added dropwise over 1 h. When the addition was finished the mixture was stirred overnight at room temperature. The solvent was evaporated under reduced pressure and the oily residue was purified by three subsequent crystallizations from first EtOAc, then Et<sub>2</sub>O and finally petroleum ether-Et<sub>2</sub>O (1 : 1) in order to remove triphenylphosphine oxide. The resulting oily mixture was purified by column chromatography on silica gel (stepwise gradient of petroleum ether-Et<sub>2</sub>O 20 : 3 to pure Et<sub>2</sub>O) to afford **5** (42.4 g, 96%) as a yellow oil; *R*<sub>f</sub> 0.54 (petroleum ether-EtOAc 10 : 1); IR (cm<sup>-1</sup>) 3076, 3040, 2927, 2860, 1772, 1708, 1641, 1614, 1468, 1459, 1441, 1396, 1372, 1334; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 7.85–7.82 (2H, m), 7.73–7.70 (2H, m), 5.74 (2H, ddt, *J* = 16.8, 10.0 and 6.8), 5.01–4.90 (4H, m), 4.26–4.18 (1H, m), 2.16–1.99 (6H, m), 1.78–1.69 (2H, m), 1.51–1.23 (4H, m); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) 169.0 (2C, s), 138.6 (2C, d), 134.1 (2C, d), 132.1 (2C, s), 123.3 (2C, d), 115.0 (2C, t), 52.0 (d), 33.4 (2C, t), 31.9 (2C, t), 26.0 (2C, t); MS (EI) *m/z* 297 (43, M<sup>+</sup>), 256 (47), 242 (25), 228 (29), 159 (40), 147 (100), 129 (63), 81 (65), 41 (76); HRMS calcd for C<sub>19</sub>H<sub>27</sub>N<sub>2</sub>O<sub>2</sub> (M + NH<sub>4</sub>) 315.2073, found 315.2072.

### 7-(1',3'-Dioxo-1',3'-dihydro-isoindol-2'-yl)-trideca-2,11-diene-2,11-dienoic acid diethyl ester (**7**)

**Method A:** Alkyl phthalimide **5** (8.751 g, 29.43 mmol) was dissolved in THF (437.6 mL) and distilled water (145.9 mL). A solution (2.5% wt in 2-methyl-2-propanol) of osmium tetroxide (5.984 g, 0.59 mmol, 2 mol%) was added in one portion. After 5 min stirring at room temperature the mixture was cooled by water bath to about 15 °C and sodium periodate (37.769 g, 176.58 mmol) was added in portions over 20 min. The mixture was stirred at room temperature for 20 h. Volatiles were evaporated under reduced pressure (bath temperature did not exceed 30 °C) to give a white solid. The solid was extracted/decanted with dichloromethane (3 × 100 mL). Combined extracts were dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated to give a crude dialdehyde **6** (10.7 g) that was used for the subsequent reaction without purification. To an oven dried flask a 60% dispersion of NaH in mineral oil (3.95 g, 98.8 mmol) was added and washed with *n*-pentane (3 × 20 mL) under argon atmosphere. The remaining solvent was evaporated under reduced pressure and freshly distilled THF (400 mL) was added. Triethyl phosphonoacetate (17.52 mL, 19.80 g, 88.3 mmol) was added dropwise over 40 min at room temperature. Stirring was continued for 2 h. The mixture was cooled to 0 °C. A solution of the crude dialdehyde **6** in freshly distilled THF (100 mL) was added dropwise over 2 h and the mixture was stirred for 48 h. A saturated solution of NH<sub>4</sub>Cl (100 mL) and ethyl acetate (100 mL) were added, the organic layer was separated and the aqueous one was extracted with ethyl acetate (3 × 200 mL). Combined organic extracts were washed with sodium bicarbonate (20 mL), brine (10 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude product (13 g) that was purified using column chromatography on silica gel (stepwise gradient of petroleum ether-ethyl acetate 9 : 1 to 1 : 1) to afford **7** (9.76 g, 75.1% over 2 steps). **Method B:** A stock solution of ruthenium(III) chloride was prepared by dissolving RuCl<sub>3</sub>·(H<sub>2</sub>O)<sub>2</sub> (842 mg, 3.5 mmol) in distilled water (100 mL) to give a concentration of 0.035 M. To a stirred solution of alkyl

phthalimide **5** (9.432 g, 31.7 mmol) and RuCl<sub>3</sub> stock solution (31.7 mL, 1.11 mmol, 3.5 mol%) in CH<sub>3</sub>CN (158.6 mL) and distilled water (26.4 mL) was added in portions NaIO<sub>4</sub> (27.2 g, 127.2 mmol) over a period of 5 min at room temperature. After completion (TLC monitoring) in 3 h, the reaction was quenched with a saturated aqueous solution of Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> (100 mL). The resulting mixture was filtered through sintered glass and the aqueous layer was extracted with AcOEt (3 × 100 mL). The combined organic layers were washed with water (50 mL), brine (50 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. The solvents were evaporated under reduced pressure to give a crude dialdehyde **6** (10.9 g) that was used for the subsequent reaction without purification. To an oven dried flask a 60% dispersion of NaH in mineral oil (7.75 g, 194.0 mmol) was added and washed with n-pentane (3 × 20 mL) under argon atmosphere. The remaining solvent was evaporated under reduced pressure and freshly distilled THF (700 mL) was added. Triethyl phosphonoacetate (31.5 mL, 35.60 g, 159.1 mmol) was added dropwise over 40 min at room temperature. Stirring was continued for 2 h. The mixture was cooled to 0 °C. A solution of the crude dialdehyde **6** in freshly distilled THF (100 mL) was added dropwise over 2 h and the mixture was stirred for 48 h. A saturated solution of NH<sub>4</sub>Cl (100 mL) and ethyl acetate (100 mL) were added, the organic layer was separated and the aqueous one was extracted with ethyl acetate (3 × 200 mL). Combined organic extracts were washed with sodium bicarbonate (20 mL), brine (10 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude product (15 g) that was purified using column chromatography on silica gel (stepwise gradient of petroleum ether–ethyl acetate 9 : 1 to 1 : 1) to afford **7** (9.9 g, 70.7% over 2 steps). **Method C:** Alkyl phthalimide **5** (1.03 g, 3.477 mmol) was dissolved in dichloromethane (20 mL). Ozone was passed through the mixture at –78 °C until a blue colour persisted. The excess of O<sub>3</sub> gas was removed by blowing argon through the mixture for 15 min. Me<sub>2</sub>S (5 mL, large excess) was added and the mixture was stirred 0.5 h at –78 °C. The cold bath was removed and stirring was continued for another 2 h. Triphenylphosphine (600 mg, 2.29 mmol) in dichloromethane (5 mL) was added and stirring was continued until no more ozonide could be detected (TLC). The mixture was concentrated under reduced pressure. The residue was triturated with Et<sub>2</sub>O (3 × 10 mL) and the combined extracts were washed with water (5 mL), brine (5 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude dialdehyde **6** (1.2 g) that was used for the subsequent reaction without purification. To an oven dried flask a 60% dispersion of NaH in mineral oil (0.51 g, 12.8 mmol) was added and washed with n-pentane (3 × 5 mL) under argon atmosphere. The remaining solvent was evaporated under reduced pressure and freshly distilled THF (40 mL) was added. Triethyl phosphonoacetate (2.07 mL, 2.339 g, 10.4 mmol) was added dropwise over 20 min at room temperature. Stirring was continued for 1 h. The mixture was cooled to 0 °C. A solution of the crude dialdehyde **6** in freshly distilled THF (10 mL) was added dropwise over 1 h and the mixture was stirred for 48 h. A saturated solution of NH<sub>4</sub>Cl (10 mL) and ethyl acetate (10 mL) were added, the organic layer was separated and the aqueous one was extracted with ethyl acetate (3 × 20 mL). Combined organic extracts were washed with sodium bicarbonate (5 mL), brine (5 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude product (1.7 g) that was purified using column chromatography on silica gel (stepwise gradient of petroleum ether–ethyl acetate 9 : 1 to 1 : 1) to afford **7** (0.688 g, 44.8% over 2 steps) as a yellow oil: *R*<sub>f</sub> 0.56 (petroleum ether–EtOAc 3 : 1); IR (cm<sup>-1</sup>) 3462, 3411, 2980, 2924, 2852, 1767, 1705, 1643, 1607, 1464, 1444, 1398, 1367; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 7.86–7.84 (2H, m), 7.78–7.75 (2H, m), 6.90 (2H, dt, *J* = 15.6 and 6.8), 5.79 (2H, d, *J* = 15.6), 4.25–4.20 (1H, m), 4.18 (4H, q, *J* = 7.2), 2.30–2.12 (6H, m), 1.80–1.72 (2H, m), 1.48–1.43 (4H, m), 1.27 (6H, t, *J* = 7.2); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 168.1 (2C, s), 165.9 (2C, s), 147.8 (2C, d), 133.6 (2C, d), 131.2 (2C, s), 122.8 (2C, d), 121.3 (2C, d), 59.6 (2C, t), 50.9 (d), 31.4 (2C, t), 31.2 (2C, t), 24.6 (2C, t), 13.8

(2C, q); MS (EI) *m/z* 441 (7, M<sup>+</sup>), 254 (13), 185 (40), 147 (100), 129 (59); HRMS calcd for C<sub>25</sub>H<sub>32</sub>NO<sub>6</sub> (M + H) 442.2230, found 442.2227.

#### 7-(2'-Hydroxymethyl-benzoylamino)-trideca-2,11-dienedioic acid diethyl ester (**10**)

To a flask containing NaBH<sub>4</sub> (0.166 g, 4.39 mmol) a solution of diester **7** (1.292 g, 2.93 mmol) in isopropanol (47.4 mL) and water (7.6 mL) was added. The mixture was stirred for 24 h at room temperature. The formation of **10** (*R*<sub>f</sub> 0.63, petroleum ether–EtOAc 1 : 1) was monitored by TLC. Brine (10 mL) was added and the mixture was extracted with Et<sub>2</sub>O (4 × 20 mL). Combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude product (1.771 g) that was purified using column chromatography on silica gel (stepwise gradient of petroleum ether–EtOAc 3 : 1 to 1 : 1) to afford **10** (0.873 g, 66.9%) as a yellow oil: IR (cm<sup>-1</sup>) 3344, 3288, 3062, 2980, 2934, 2852, 1716, 1629, 1536, 1454, 1362, 1311, 1260, 1178; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 7.54–7.34 (4H, m), 6.96–6.88 (2H, m), 6.51–6.39 (1H, m), 5.82 (2H, dd, *J* = 15.6 and 1.2), 4.57 (2H, s), 4.20–4.10 (5H, m), 2.31–2.22 (4H, m), 2.0 (1H, bs), 1.63–1.31 (8H, m), 1.30–1.23 (6H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 173.8 (s), 169.7 (s), 166.7 (s), 148.5 (2C, d), 139.6 (s), 136.0 (s), 131.1 (d), 130.7 (d), 128.2 (d), 127.7 (d), 121.8 (2C, d), 64.6 (t), 60.3 (2C, t), 49.4 (d), 34.8 (2C, t), 31.8 (2C, t), 24.5 (2C, t), 14.3 (2C, q); MS (EI) *m/z* 445 (21, M<sup>+</sup>), 428 (100), 427 (48), 416 (25), 135 (75), 132 (99); HRMS calcd for C<sub>25</sub>H<sub>36</sub>NO<sub>6</sub> (M + H) 446.2542, found 446.2543.

#### (4S\*, 6S\*)-(6-Ethoxycarbonylmethyl-octahydro-quinolizin-4-yl)-acetic acid ethyl ester (**8**)

To a flask containing NaBH<sub>4</sub> (1.046 g, 27.65 mmol) a solution of diester **7** (8.721 g, 19.75 mmol) in isopropanol (320.0 mL) and water (51.6 mL) was added. The mixture was stirred for 24 h at room temperature. The formation of **10** was monitored by TLC. Acetone (5.23 mL, 71.23 mmol) was added and stirring was continued overnight. The mixture was cooled to 0 °C and glacial AcOH (33.9 mL, 592.76 mmol) was added dropwise. The mixture was heated to 80 °C (bath temperature) under stirring for 24 h. The mixture was allowed to cool and brine (20 mL) was added. Solids were filtered off and petroleum ether (100 mL) was added to the filtrate to initiate separation of organic and aqueous layers. The aqueous layer was extracted with ethyl acetate (3 × 100 mL). Combined organic layers were washed with brine (20 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude hydroacetate of quinolizidine **8** that was taken into benzene (200 mL) and K<sub>2</sub>CO<sub>3</sub> (5.6 g) was added. The mixture was stirred overnight. Solids were filtered off and the filtrate was evaporated. The residue was crystallized from first Et<sub>2</sub>O and then from Et<sub>2</sub>O–petroleum ether 1 : 1 to separate phthalide **11** (*R*<sub>f</sub> 0.48, alumina, petroleum ether–EtOAc 5 : 1). Crystals of phthalide **11** were filtered off, washed with Et<sub>2</sub>O–petroleum ether 1 : 1 (3 × 2 mL) and the combined filtrates were evaporated to give crude free base **8** (5.3 g) that was purified using column chromatography on activity IV neutral alumina (40 g, petroleum ether–EtOAc 30 : 1) to afford **8** (4.38 g, 71.2%) as a yellow oil: *R*<sub>f</sub> 0.79 (alumina, petroleum ether–EtOAc 5 : 1); IR (cm<sup>-1</sup>) 2980, 2929, 2853, 1726, 1465, 1445, 1368, 1337, 1301, 1276, 1163, 1096, 1030; <sup>1</sup>H NMR (400 MHz, benzene-d<sub>6</sub>) 4.04–3.93 (4H, m), 3.83–3.80 (1H, m), 2.81 (1H, dd, *J* = 14.2 and 4.4), 2.78–2.72 (1H, m), 2.65 (1H, dd, *J* = 14.6 and 3.2), 2.41 (1H, dd, *J* = 14.6 and 10.4), 2.29 (1H, dd, *J* = 14.2 and 6.8), 1.98–1.92 (1H, m), 1.72–1.66 (3H, m), 1.47–1.36 (2H, m), 1.34–1.29 (4H, m), 1.24–1.04 (3H, m), 1.02–0.98 (6H, m); <sup>13</sup>C NMR (75 MHz, benzene-d<sub>6</sub>) 171.9 (s), 170.9 (s), 59.0 (t), 58.9 (t), 53.9 (d), 52.9 (d), 50.0 (d), 38.5 (t), 33.7 (t), 33.6 (t), 32.0 (t), 28.6 (t), 26.3 (t), 22.9 (t), 17.7 (t), 12.9 (2C, q); MS (EI) *m/z* 311 (60, M<sup>+</sup>), 224 (100), 167 (57), 149 (95), 136 (62); HRMS calcd for C<sub>17</sub>H<sub>30</sub>NO<sub>4</sub> (M + H) 312.2175, found 312.2171.

**(4*S*\*, 6*S*\*)-(6-Ethoxycarbonylmethyl-octahydro-quinolizin-4-yl)-acetic acid ethyl ester (8), (2'*S*\*, 6'*S*\*)-6-(6'-ethoxycarbonylmethyl-piperidin-2'-yl)-hexanoic acid ethyl ester (13) and 7-amino-tridecanedioic acid diethyl ester (14)**

To a flask containing NaBH<sub>4</sub> (261 g, 6.9 mmol) a solution of diester **7** (2.437 g, 5.52 mmol) in isopropanol (89.4 mL) and water (14.4 mL) was added. The mixture was stirred for 24 h at room temperature. The formation of **10** was monitored by TLC. The mixture was cooled to 0 °C and glacial AcOH (9.46 mL, 165.4 mmol) was added dropwise. The mixture was heated to 80 °C (bath temperature) under stirring for 24 h. The mixture was allowed to cool down and brine (10 mL) was added. Solids were filtered off and petroleum ether (10 mL) was added to the filtrate to initiate separation of organic and aqueous layers. The aqueous layer was extracted with ethyl acetate (3 × 100 mL). Combined organic layers were washed with brine (5 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a mixture of crude amine hydroacetates that was taken into benzene (50 mL) and K<sub>2</sub>CO<sub>3</sub> (3 g) was added. The mixture was stirred overnight. Solids were filtered off and the filtrate was evaporated. The residue was crystallized from first Et<sub>2</sub>O and then from Et<sub>2</sub>O–petroleum ether 1 : 1 to separate phthalide **11**. Crystals of phthalide **11** were filtered off, washed with Et<sub>2</sub>O–petroleum ether 1 : 1 (3 × 2 mL) and the combined filtrates were evaporated to give a mixture of crude free amines (2.5 g) that was purified using column chromatography on activity IV neutral alumina (40 g, stepwise gradient petroleum ether–EtOAc 30 : 1 to MeOH pure) to afford **8** (0.539 g, 31.4%), **13** (0.36 g, 21.0%) and **14** (0.275 g, 15.8%) as yellow oils.

For **13**: *R*<sub>f</sub> 0.30 (alumina, petroleum ether–EtOAc 5 : 1); IR (cm<sup>-1</sup>) 3385, 2929, 2852, 1731, 1454, 1372, 1183; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 4.17–4.10 (4H, m), 2.98–2.90 (1H, m), 2.56–2.50 (1H, m), 2.41 (2H, d, *J* = 6.0), 2.29 (2H, t, *J* = 7.6), 2.20 (1H, bs), 1.81–1.76 (2H, m), 1.66–1.58 (6H, m), 1.50–1.30 (6H, m), 1.28–1.24 (6H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 173.8 (s), 172.6 (s), 60.4 (t), 60.2 (t), 56.7 (d), 53.4 (d), 41.5 (t), 37.1 (t), 34.3 (t), 32.5 (t), 32.1 (t), 29.3 (t), 25.6 (t), 24.9 (t), 24.6 (t), 14.2 (2C, q); MS (EI) *m/z* 313 (2, M<sup>+</sup>), 300 (99), 290 (20), 276 (26), 226 (15), 156 (100), 124 (29), 82 (44); HRMS calcd for C<sub>17</sub>H<sub>32</sub>NO<sub>4</sub> (M + H) 314.2326, found 314.2324.

For **14**: *R*<sub>f</sub> 0.07 (alumina, petroleum ether–EtOAc 5 : 1); IR (nujol, cm<sup>-1</sup>) 3426, 2955, 2852, 1726, 1577; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>) 4.12 (4H, q, *J* = 7.2), 3.02–3.19 (1H, m), 2.30 (4H, t, *J* = 7.1), 1.98–1.32 (16H, m), 1.25 (6H, t, *J* = 7.2); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 173.6 (2C, s), 60.3 (2C, t), 52.2 (d), 34.2 (2C, t), 33.0 (2C, t), 28.8 (2C, t), 25.1 (2C, t), 24.7 (2C, t), 14.3 (2C, q); MS (EI) *m/z* 315 (1, M<sup>+</sup>), 184 (25), 170 (100), 82 (38); HRMS calcd for C<sub>17</sub>H<sub>34</sub>NO<sub>4</sub> (M + H) 316.2488, found 316.2492.

**(3*aS*\*, 9*aS*\*)-2-Oxo-dodecahydro-pyrido[2,1,6-*de*]quinolizine (18)**

**Method A:** An oven dried double-neck flask was equipped with magnetic stirring bar and a Dean–Stark trap. The flask was charged with *t*BuOK (3.899 g, 34.74 mmol) and absolute benzene (180 mL). The Dean–Stark trap was filled with absolute benzene and charged with a lithium stick (1 g). Quinolizidine **8** (5.41 g, 17.37 mmol) was added in absolute benzene (20 mL) *via* a septum. The mixture was refluxed (bath temperature 120 °C) under continuous removal of EtOH for 2 h. The septum was replaced by a glass stopper and reflux was continued for another 17 h. The formation of **16** (*R*<sub>f</sub> 0.46, MeOH–EtOAc 1 : 10) and **17** (*R*<sub>f</sub> 0.33, MeOH–EtOAc 1 : 10) was monitored by TLC. The mixture was allowed to cool down and water (50 mL) was added. The organic layer was separated and the aqueous layer was extracted with EtOAc (3 × 100 mL). Combined organic layers were washed with brine (25 mL) and finally dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude mixture of **16** and **17** (4.1 g) that was purified by filtration through a pad of activity IV neutral alumina (2 g, EtOAc) to afford a regioisomeric mixture of keto

and enol tautomers **16** and **17** (3.85 g, 83.5%) as a yellow oil. The mixture of **16** and **17** (1.6 g, 6.03 mmol) was dissolved in CHCl<sub>3</sub> (10 mL) and 2 M HCl<sub>g</sub> in Et<sub>2</sub>O (40 mL) was added. The mixture was stirred for 1 h at room temperature. The volatiles were evaporated under reduced pressure and the residue was taken into DMF (50 mL). Water (540 μL, 30.15 mmol) and LiCl (1.278, 30.15 mmol) were added. The mixture was heated at reflux (oil bath 140 °C) for 4 h. The volatiles were evaporated at 60 °C under reduced pressure. The residue was taken into dichloromethane (20 mL) and filtered through a pad of activity IV neutral alumina (2 g, dichloromethane). Evaporation gave a crude product (0.97 g) that was purified using column chromatography on activity IV neutral alumina (10 g, stepwise gradient of hexane–Et<sub>2</sub>O 20 : 1 to 1 : 1) to afford **18** (0.764 g, 65.6%) as a yellow oil. **Method B:** To a stirred solution of quinolizidine **8** (108.2 mg, 0.347 mmol) in absolute THF (10 mL) was added LDA (74.3 mg, 0.694 mmol) *via* septum at –78 °C and under argon atmosphere. The formation of **16** and **17** was monitored by TLC. After 8 h the mixture was allowed to reach room temperature. THF was evaporated and the residue was taken into benzene (10 mL). Water (10 mL) was added. The organic layer was separated and the aqueous one was extracted with EtOAc (3 × 10 mL). Combined organic layers were dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude mixture of **16** and **17** (100 mg) that was purified by filtration through a pad of activity IV neutral alumina (2 g, EtOAc) to afford a mixture of pure **16** and **17** (44.3 mg, 48.1%) as a yellow oil. This mixture was directly converted to ketone **18** in the same way as in Method A. For **18**: *R*<sub>f</sub> 0.49 (alumina, petroleum ether–EtOAc 1 : 1); IR (cm<sup>-1</sup>) 3422, 2932, 2853, 1715, 1460, 1442, 1367, 1340, 1288; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 3.43–3.38 (1H, m), 3.23–3.13 (2H, m), 2.89 (1H, dd, *J* = 13.6 and 6.8), 2.26–2.14 (3H, m), 1.93–1.82 (3H, m), 1.72–1.67 (1H, m), 1.61–1.48 (5H, m), 1.46–1.00 (3H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 209.1 (s), 61.0 (d), 57.9 (d), 48.9 (d), 48.7 (t), 47.5 (t), 34.9 (t), 30.9 (t), 25.6 (t), 23.8 (t), 21.6 (t), 18.6 (t); MS (EI) *m/z* 193 (49, M<sup>+</sup>), 178 (22), 150 (100), 122 (80), 82 (42); HRMS calcd for C<sub>12</sub>H<sub>20</sub>NO (M + H) 194.1539, found 194.1535.

**(3*aS*\*, 9*aS*\*)-2-Methylene-dodecahydro-pyrido[2,1,6-*de*]quinolizine (3)**

To a stirred suspension of methyltriphenylphosphonium bromide (1.0314 g, 2.89 mmol) in absolute THF (69 mL) was added 2.5M *n*BuLi in hexanes (1 mL, 2.5 mmol) at –78 °C and under argon atmosphere. When the addition was complete, the cold bath was removed and the yellow mixture was stirred for 75 min. A solution of ketone **18** (382 mg, 1.98 mmol) in absolute THF (10 mL) was added over 5 min. The mixture was stirred at room temperature for 1 h and then heated to 60 °C for 30 min. The mixture was allowed to cool down and water (10 mL) was added. The mixture was extracted with EtOAc (3 × 30 mL). Combined extracts were washed with brine (2 × 10 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a mixture (0.8 g) that was purified by crystallization from EtOAc and then from petroleum ether–Et<sub>2</sub>O 1 : 1 to get rid of Ph<sub>3</sub>PO. The remaining oil (0.4 g) was further purified using column chromatography on activity IV neutral alumina (71 g, hexane–Et<sub>2</sub>O 500 : 75) to afford **3** (330.45 g, 87.4%) as a yellow oil: *R*<sub>f</sub> 0.59 (alumina, MeOH–EtOAc 1 : 1); IR (cm<sup>-1</sup>) 3067, 2924, 2847, 1737, 1650, 1460, 1440; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 4.78–4.59 (2H, m), 3.18–3.05 (2H, m), 3.00–2.86 (1H, m), 2.72–2.56 (1H, m), 2.22–1.26 (1H, m), 2.12–2.02 (1H, m), 2.00–1.85 (5H, m), 1.78–1.44 (5H, m), 1.36–1.18 (1H, m), 1.12–0.92 (2H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 144.3 (s), 108.3 (t), 59.6 (d), 58.2 (d), 49.4 (d), 42.7 (t), 40.5 (t), 34.5 (t), 31.1 (t), 25.7 (t), 22.6 (t), 21.9 (t), 19.4 (t); MS (EI) *m/z* 191 (61, M<sup>+</sup>), 176 (52), 162 (38), 148 (100), 136 (82), 120 (52), 94 (57), 79 (57); HRMS calcd for C<sub>13</sub>H<sub>22</sub>N (M + H) 192.1747, found 192.1749.



### (±)-Hippodamine (1)

Mesitylenesulfonic acid dihydrate (245.3 mg, 1.04 mmol) was dissolved in THF (25 mL). To this solution olefin **3** (99.3 mg, 0.52 mmol) in THF (25 mL) was added in one portion. The mixture was allowed to stir at room temperature for 1 h. Then 10% palladium on carbon (100 mg) was added and hydrogen (normal pressure) was applied. The formation of **1** ( $R_f$  0.46, alumina, MeOH–EtOAc 1 : 1) was monitored by TLC after a mini work-up. After 2 days the solids were filtered off and THF evaporated. The residue was taken into benzene and  $K_2CO_3$  (1 g) was added. The mixture was stirred overnight. Solids were filtered off and the filtrate was evaporated. The remaining oil (95 mg) was further purified using column chromatography on activity IV neutral alumina (2.3 g, hexane–Et<sub>2</sub>O 5 : 1) to afford a 3.5 : 1 mixture (by <sup>1</sup>H NMR) of **1** and **2** (93.3 mg, 93.0%). The mixture was further subjected to a multiple column chromatography (4 times) on activity IV neutral alumina (12 g, hexane–Et<sub>2</sub>O 10 : 1, sample applied in a small amount of chlorobenzene) to finally afford isomerically pure hippodamine (**1**) (37.1 mg, 37.0%);  $R_f$  0.46 (alumina, MeOH–EtOAc 1 : 1); IR (cm<sup>-1</sup>) 2920, 2859, 1455, 1443, 1294, 1132, 1102; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 3.00–2.95 (2H, m), 2.83 (1H, dddd,  $J = 11.2, 11.2, 2.8, 2.8$ ), 1.93–1.77 (4H, m), 1.70–1.63 (1H, m), 1.59–1.43 (8H, m), 1.21–1.12 (1H, m), 1.04–0.94 (2H, m), 0.87–0.78 (1H, m), 0.85 (3H, d,  $J = 6.4$ ); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 58.8 (d), 58.1 (d), 47.8 (d), 43.4 (t), 40.0 (t), 34.6 (t), 31.4 (t), 26.1 (t), 25.7 (d), 23.1 (t), 22.5 (q), 22.3 (t), 19.7 (t); MS (EI)  $m/z$  193 (22, M<sup>+</sup>), 192 (42), 178 (18), 164 (26), 151 (35), 150 (45), 137 (28), 136 (42), 105 (25), 77 (35), 67 (31), 55 (54), 43 (43), 41 (100); HRMS calcd for C<sub>13</sub>H<sub>24</sub>N (M + H) 194.1903, found 194.1902. Synthetic hippodamine proved identical to the natural material by IR, <sup>1</sup>H NMR, <sup>13</sup>C NMR, and MS spectral comparison.<sup>26,27</sup>

### (2*R*\*, 3*aS*\*, 6*aS*\*, 9*aS*\*)-2-Hydroxymethyl-dodecahydro-pyrido[2,1,6-*de*]quinolizine (22)

Olefin **3** (53.4 mg, 0.279 mmol) was dissolved in absolute THF (20 mL) and the solution was stirred at –78 °C under argon atmosphere. A 0.5 M THF solution of 9-borabicyclo[3.3.1]nonane-THF (9-BBN·THF) (136.1 mg, 1.116 mmol) was added dropwise *via* a septum. Stirring was continued for 1 h at –78 °C. Then the mixture was allowed to reach room temperature and stirring was continued overnight. The mixture was then refluxed for 2 h. The mixture was allowed to reach room temperature and water (1 mL) was added followed by MeOH (9.4 mL) and 3 N NaOH (0.93 mL, 1.116 mmol). The mixture was cooled to 0 °C and 11.6 M (or 30%) H<sub>2</sub>O<sub>2</sub> (241 μL, 2.79 mmol) was added dropwise over 10 min. The mixture was stirred for 2 h. A white precipitate was filtered off. The filtrate was evaporated under reduced pressure and the residue was taken into Et<sub>2</sub>O (10 mL). This solution was washed with water (5 mL), brine (5 mL) and finally dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude product (97 mg) that was purified using column chromatography on activity IV neutral alumina (5 g, stepwise gradient hexane–Et<sub>2</sub>O 5 : 1, pure Et<sub>2</sub>O to pure MeOH) to afford **22** (54 mg) as a white solid. The solid was further crystallized from Et<sub>2</sub>O to give **22** (49.3 mg, 84.4%) as crystals (clear, colourless, tapered rectangular prisms);  $R_f$  0.32 (alumina, MeOH–EtOAc 1 : 1); mp 145.5–146.0 °C (Et<sub>2</sub>O); IR (cm<sup>-1</sup>) 3652, 3354, 2930, 2848, 1731, 1644, 1582, 1460, 1440; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 3.79 (1H, bt,  $J = 9.6$ ), 3.63 (1H, bdd,  $J = 9.6$  and 6.0), 2.96–2.84 (3H, m), 2.14–2.07 (1H, m), 1.95–1.77 (4H, m), 1.67–1.38 (9H, m), 1.28–1.13 (2H, m), 1.06–1.00 (1H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 66.5 (t), 58.2 (d), 57.8 (d), 43.6 (d), 35.1 (t), 34.8 (d), 34.6 (t), 32.0 (t), 31.4 (t), 26.7 (t), 26.4 (t), 23.2 (t), 19.7 (t); MS (EI)  $m/z$  209 (38, M<sup>+</sup>), 208 (100), 178 (69), 167 (47), 150 (55), 137 (40); HRMS calcd for C<sub>13</sub>H<sub>23</sub>NO (M + H) 210.1852, found 210.1855. Anal. Calcd. for C<sub>13</sub>H<sub>23</sub>NO: C, 74.59; H, 11.07; N, 6.69. Found: C, 74.65; H, 11.13; N, 6.38%.

### (2*S*\*, 3*aS*\*, 6*aR*\*, 9*aS*\*)-2-Hydroxymethyl-dodecahydro-pyrido[2,1,6-*de*]quinolizine (21) and (2*R*\*, 3*aS*\*, 6*aS*\*, 9*aS*\*)-2-hydroxymethyl-dodecahydro-pyrido[2,1,6-*de*]quinolizine (22)

Olefin **3** (10.0 mg, 0.052 mmol) was dissolved in absolute THF (10 mL) and the solution was stirred at –78 °C under argon atmosphere. A 1 M THF solution of borane-THF (210 μL, 0.210 mmol) was added dropwise *via* a septum. Stirring was continued for 1 h at –78 °C. Then the mixture was allowed to reach room temperature and stirring was continued overnight. The mixture was then refluxed for 4 h. The mixture was allowed to reach room temperature and water (11.2 μL, 0.624 mmol) was added followed by MeOH (109 μL) and 3 N NaOH (208 μL, 0.624 mmol). The mixture was cooled to 0 °C and 11.6 M (or 30%) H<sub>2</sub>O<sub>2</sub> (44.9 μL, 0.520 mmol) was added dropwise over 10 min. The mixture was stirred for 2 h. A white precipitate was filtered off. The filtrate was evaporated under reduced pressure and the residue was taken into Et<sub>2</sub>O (10 mL). This solution was washed with water (5 mL), brine (5 mL) and finally dried over Na<sub>2</sub>SO<sub>4</sub>. Evaporation gave a crude product (20 mg) that was purified using column chromatography on activity IV neutral alumina (2.5 g, stepwise gradient hexane–Et<sub>2</sub>O 5 : 1, pure Et<sub>2</sub>O to pure MeOH) to afford a white solid. According to <sup>1</sup>H NMR the solid was a mixture of **21** and **22** in a ratio 1 : 4.25 (9.81 mg, 90.0%).

### (2*R*\*, 3*aS*\*, 6*aS*\*, 9*aS*\*)-2-(*p*-Toluenesulfonyloxymethyl)-dodecahydro-pyrido[2,1,6-*de*]quinolizine (23)

A solution of alcohol **22** (10 mg, 0.05 mmol) and triphenylmethane (2 mg) in absolute THF (10 mL) was cooled to 0 °C under stirring and argon atmosphere. A 2.5 M solution of *n*BuLi in hexanes (21 μL, 0.053 mmol) was added from a syringe *via* a septum until the solution became slightly pink. The precipitated lithium alkoxide was stirred for 10 min at 0 °C. Then *p*-toluenesulfonyl chloride (9.1 mg, 0.05 mmol) in absolute THF (5 mL) was added dropwise. The resulting solution was stirred for 2.5 h at 0 °C. The formation of **23** was monitored by TLC. The mixture was diluted with Et<sub>2</sub>O (10 mL), washed with water (2 mL), brine (2 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. The volatiles were evaporated to give a crude product (30 mg) that was purified using column chromatography on activity IV neutral alumina (15 g, stepwise gradient of hexane–Et<sub>2</sub>O 5 : 1, 1 : 1 to 1 : 5) to afford **23** (16.5 mg, 94.8%) as a yellow oil; IR (cm<sup>-1</sup>) 2925, 2853, 1598, 1460, 1440, 1363, 1255, 1178, 1196; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 7.82–7.80 (2H, m), 7.38–7.36 (2H, m), 4.20 (1H, t,  $J = 9.6$ ), 3.99 (1H, dd,  $J = 9.6$  and 6.0), 2.93–2.86 (2H, m), 2.63–2.57 (1H, m), 2.46 (3H, s), 2.17–2.02 (2H, m), 1.82–1.66 (3H, m), 1.51–1.20 (9H, m), 1.18–0.78 (3H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 144.8 (s), 133.0 (s), 129.8 (2C, d), 128.0 (2C, d), 73.3 (t), 58.1 (d), 57.3 (d), 43.2 (d), 34.5 (t), 34.4 (t), 31.5 (t), 31.2 (t), 31.0 (d), 26.6 (t), 26.3 (t), 23.0 (t), 21.7 (q), 19.5 (t); MS (EI)  $m/z$  363 (6, M<sup>+</sup>), 208 (100), 188 (50), 176 (51), 148 (50), 91 (99); HRMS calcd for C<sub>20</sub>H<sub>30</sub>NO<sub>3</sub>S (M + H) 364.1941, found 364.1942.

### *epi*-(±)-Hippodamine (2)

To a solution of tosylate **23** (26.75 mg, 0.0736 mmol) in 2-butanone (20 mL) was added sodium iodide (110.21 mg, 0.735 mmol). The mixture was refluxed for 24 h under argon atmosphere and exclusion of light. The mixture was evaporated to dryness. Water (5 mL) was added and the resulting solution was extracted with EtOAc (3 × 10 mL). Combined extracts were washed with a saturated solution of sodium thiosulfate (5 mL), brine (2 mL) and dried over Na<sub>2</sub>SO<sub>4</sub>. The volatiles were evaporated to give crude **24** (23.4 mg) as a yellow oil ( $R_f$  0.45, alumina, MeOH–EtOAc 1 : 1) that was used without further purification. A 5 mL round-bottom flask was charged with activated zinc<sup>25</sup> (0.2 g, 3.06 mmol) and glacial acetic acid (3 mL) was added. The mixture was exposed to sonication for 5 min. Then crude iodide **24** (~23.4 mg, 73.3 μmol) was added

in glacial acetic acid (0.5 mL). The mixture was refluxed for 3 h. The mixture was allowed to reach room temperature and then filtered. The solids were washed with MeOH (1 mL) and the filtrate was evaporated. The residue was diluted with benzene (10 mL) and  $K_2CO_3$  (1 g) was added. The mixture was stirred for 4 h. Solids were filtered off and the filtrate was evaporated to dryness to give a crude product (22 mg) that was purified using column chromatography on activity IV neutral alumina (1 g, hexane–Et<sub>2</sub>O 4 : 1) to afford **29** (11.8 mg, 83.1% over two steps) as a colourless oil:  $R_f$  0.40 (alumina, MeOH–EtOAc 1 : 1); IR (cm<sup>-1</sup>) 2925, 2853, 1455, 1440, 1378, 1265, 1122, 1075, 1024; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 2.97–2.88 (3H, m), 2.13–2.06 (1H, m), 1.98–1.92 (1H, m), 1.89–1.79 (4H, m), 1.52–1.15 (10H, m), 1.10 (3H, d,  $J = 7.2$ ), 1.03–0.9 (1H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 58.4 (d), 58.3 (d), 43.1 (d), 40.5 (t), 36.6 (t), 34.7 (t), 31.5 (t), 27.0 (t), 26.7 (t), 25.7 (d), 23.4 (t), 22.1 (q), 19.8 (t); MS (EI)  $m/z$  193 (54, M<sup>+</sup>), 192 (100), 178 (57), 165 (54), 151 (60), 150 (71), 111 (97); HRMS calcd for C<sub>13</sub>H<sub>24</sub>N (M + H) 194.1903, found 194.1905.

#### (2R\*, 3aS\*, 6aS\*, 9aS\*)-2-Hydroxy-2-methyl-dodecahydro-pyrido[2,1,6-de]quinolizine (**25**)

To a stirred solution of ketone **18** (87.2 mg, 0.451 mmol) in toluene (10 mL) cooled to 0 °C under argon atmosphere was added 1.6 M MeLi in Et<sub>2</sub>O (846.0 μL, 1.353 mmol) in three portions over 3 h. Stirring was continued for another 5 h. A saturated solution of NH<sub>4</sub>Cl (1 mL) was added. The organic layer was separated and the aqueous one was extracted with benzene (3 × 5 mL). Combined extracts were dried over Na<sub>2</sub>SO<sub>4</sub>. The volatiles were evaporated to give a crude product (67 mg) that was purified using column chromatography on activity IV neutral alumina (10.6 g, stepwise gradient of hexane–Et<sub>2</sub>O 5 : 1 to 1 : 1 and finally pure Et<sub>2</sub>O). The product was allowed to crystallize from Et<sub>2</sub>O to afford **25** (57.8 mg, 61.2%) as crystals (clear, colourless plates):  $R_f$  0.38 (alumina, MeOH–EtOAc 1 : 1); mp 146.8–147.1 °C (Et<sub>2</sub>O); IR (cm<sup>-1</sup>) 3365, 2919, 2854, 1459, 1442; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) 3.14–3.06 (1H, m), 3.02–2.94 (2H, m), 2.36–2.24 (1H, m), 1.96–1.76 (4H, m), 1.64–1.40 (8H, m), 1.36–1.24 (1H, m), 1.20 (3H, s), 1.17–0.99 (2H, m); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) 69.2 (s), 58.2 (d), 58.1 (d), 47.4 (t), 43.7 (d), 42.8 (t), 34.3 (t), 32.7 (q), 31.5 (t), 26.6 (t), 25.1 (t), 23.0 (t), 19.5 (t); MS (EI)  $m/z$  209 (64, M<sup>+</sup>), 208 (100), 190 (42), 176 (35), 167 (61), 150 (66), 55 (54), 43 (90), 41 (99); HRMS calcd for C<sub>15</sub>H<sub>24</sub>NO (M + H) 210.1852, found 210.1851.

#### Deoxygenation of **25**

To a solution of tertiary alcohol **25** (23.5 mg, 0.112 mmol) in absolute THF (10 mL) 2.5 M *n*BuLi in hexanes (49.3 μL, 0.123 mmol) was added dropwise under stirring, argon atmosphere and at –78 °C. After 1 h carbon disulfide (26.9 μL, 0.448 mmol) was added and the mixture was allowed to reach room temperature. Stirring was continued for 3 h. MeI (27.9 μL, 0.448 mmol) was added and the mixture was stirred overnight. The mixture was filtered. The filtrate was evaporated and filtered again through a short column of neutral alumina activity IV. The crude xanthate **26** (14.0 mg) was used without any further purification. To a refluxing solution of tri-*n*-butyltin hydride (27.4 mg, 0.094 mmol) and AIBN (0.8 mg, 4.7 μmol) in toluene (10 mL) was added a solution of xanthate **26** (14.0 mg, 0.047 mmol) in toluene (7 mL) dropwise over 1 h. The mixture was refluxed for a further 4 h and left to stand overnight. Neutral alumina activity IV (0.5 g) was added and the mixture was evaporated to dryness under reduced pressure. The solid was purified using column chromatography on activity IV neutral alumina (10.0 g, hexane–Et<sub>2</sub>O 10 : 1) to give a mixture of **1** and **2** (combined yield 4.1 mg, 19% over 2 steps) in a ratio 2.3 : 1 (by <sup>1</sup>H NMR).

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- 22 Compound **25**. Crystal data: C<sub>15</sub>H<sub>23</sub>NO,  $M = 209.3$ . Monoclinic, space group  $P2_1/c$  (no. 14),  $a = 9.766(3)$ ,  $b = 9.337(2)$ ,  $c = 13.767(2)$  Å,  $b = 102.38(2)^\circ$ ,  $V = 1226.1(5)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 1.134$  g cm<sup>-3</sup>,  $F(000) = 464$ ,  $T = 293(2)$  K,  $\mu(\text{Mo-K}\alpha) = 0.7$  cm<sup>-1</sup>,  $\lambda(\text{Mo-K}\alpha) = 0.71069$  Å. A clear, colourless plate crystal,  $ca. 0.62 \times 0.21 \times 0.05$  mm, was mounted on a Nonius CAD4 diffractometer; of the 2511 reflections measured to  $\theta_{\text{max}} = 25^\circ$ , 2146 were unique ( $R_{\text{int}} = 0.012$ ) and 1263 'observed' with  $I > 2\sigma_I$ . Refinement gave  $wR_2 = 0.136$  and  $R_1 = 0.081$ <sup>29</sup> for all 2146 reflections weighted  $w = [\sigma^2(F_o^2) + (0.0645P)^2]^{-1}$ ; for the 'observed' data only,  $R_1 = 0.047$ . CCDC reference number 248862. See <http://www.rsc.org/suppdata/ob/b4/b413052a/> for crystallographic data in .cif or other electronic format.
- 23 Compound **22**. Crystal data: C<sub>15</sub>H<sub>23</sub>NO,  $M = 209.3$ . Monoclinic, space group  $P2_1/n$  (equiv. to no. 14),  $a = 7.660(2)$ ,  $b = 14.610(3)$ ,  $c = 10.367(2)$  Å,  $b = 91.13(3)^\circ$ ,  $V = 1160.0(4)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 1.199$  g cm<sup>-3</sup>,  $F(000) = 464$ ,  $T = 140(1)$  K,  $\mu(\text{Mo-K}\alpha) = 0.7$  cm<sup>-1</sup>,  $\lambda(\text{Mo-K}\alpha) = 0.71069$  Å. A clear, colourless tapered, rectangular prismatic crystal,  $ca. 0.50 \times 0.25 \times 0.20$  mm, was mounted on a Rigaku R-Axis IIC image-plate diffractometer. Total no. of reflections recorded to  $\theta_{\text{max}} = 25.4^\circ$ , was 6174 of which 2043 were unique ( $R_{\text{int}} = 0.052$ ); 1851 were 'observed' with  $I > 2\sigma_I$ . Refinement gave  $wR_2 = 0.103$  and  $R_1 = 0.042$ <sup>29</sup> for all 2043 reflections weighted  $w = [\sigma^2(F_o^2) + (0.0384P)^2 + 0.38P]^{-1}$  with  $P = (F_o^2 + 2F_c^2)/3$ ; for the 'observed' data only,  $R_1 = 0.039$ . CCDC reference number 248861. See <http://www.rsc.org/suppdata/ob/b4/b413052a/> for crystallographic data in .cif or other electronic format.
- 24 Tosylation of alcohol **22** using tosyl chloride, triethyl amine and trimethylammonium hydrochloride in dichloromethane gave a low yield of **23** (Method A in Scheme 7). For this reason **22** was first deprotonated using *n*-BuLi and then tosyl chloride was added giving

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- the desired tosylate **23** in 95% yield (Method B in Scheme 7).  
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